

Safety Data Sheet

Section 01 - Identification

Product Identifier Sodium Nitrate

Other Means of Identification Nitratine, nitric acid, sodium salt, sodium saltpeter, sodium nitrate, crystal.

Product Use and Restrictions on

Use

Laboratory reagent

Initial Supplier Identifier ClearTech Industries Inc.

1500 Quebec Avenue Saskatoon, SK. Canada

S7K 1V7

Prepared By ClearTech Industries Inc. Technical Writer

Phone: 1 (800) 387-7503

24-Hour Emergency Phone Phone: 1 (306) 664 – 2522

Section 02 - Hazard Identification

GHS-Classification

Eye Damage/Irritation Category 2

Physical Hazards

Oxidizing Solid Category 3

Warning

Hazard Statements

H272 – May intensify fire; oxidiser. H319 – Causes serious eye irritation.

Pictograms



Precautionary Statements

P210 - Keep away from heat, sparks, open flames, and hot surfaces. — No smoking.

P220 – Keep/Store away from clothing and combustible materials.

P280 – Wear eye protection and face protection.

P370 + P378 – In case of fire: Use extinguishing media suitable for surrounding fire.

P264 – Wash hands thoroughly after handling.

P305 + P351 + P338 - IF IN EYES: Rinse cautiously with water for several minutes. Remove contact lenses, if present and easy to do. Continue rinsing.

P337 + P313 - If eye irritation persists: Get medical advice/attention.

P501 – Dispose of contents/container in accordance with all federal, provincial, and/or local regulations including the Canadian Environmental Protection Act.

Section 03 - Composition / Information on Ingredients

Chemical Name CAS Number Weight % **Unique Identifiers**

Sodium Nitrate 7631-99-4 98-100%

Section 04 - First Aid Measures

If symptoms are experienced, remove victim to fresh air. Give artificial respiration only if Inhalation

breathing has stopped. If breathing is difficult, give oxygen. Seek medical attention.

Rinse skin with lukewarm, gently flowing water and non-abrasive soap. Remove **Skin Contact / Absorption**

contaminated clothing. Seek medical attention if irritation occurs or persists.

Eye Contact Immediately flush eye(s) with lukewarm, gently flowing water for 30 minutes while forcibly

holding the eyelids open to ensure complete irrigation of the eye tissue. If a contact lens is

present, remove only if easy to do so. If irritation persists, seek medical attention.

Do NOT induce vomiting. Never give anything by mouth to an unconscious or convulsing Ingestion

person. Rinse mouth out with water. Seek medical attention. If vomiting occurs spontaneously, keep head below hips to prevent aspiration of liquid into the lungs.

Additional Information Treatment based on sound judgment of physician and individual reactions of patient.

Section 05 - Fire Fighting Measures

Sodium nitrate is not combustible. Use extinguishing agents compatible with sodium Suitable Extinguishing Media

> nitrate and appropriate for the surrounding fire. Use flooding quantities of water to fight fires in which this material is involved. Note that large quantities of sodium nitrate may melt or fuse in a fire and that water application may result in extensive scattering of

molten material.

Do not use dry chemical extinguishing agent on oxidizer fires. Carbon dioxide or other **Unsuitable Extinguishing Media**

extinguishing agents that smother flames are not effective in fires involving oxidizers.

Specific Hazards Arising From the

Chemical

Mild oxidizer – slightly increases the burning rate of combustible materials. Prolonged contact with combustible materials may produce enough heat to ignite the combustible material. Appreciable decomposition of sodium nitrate begins at 510-600°C forming sodium nitrite and oxygen. In a fire, sodium nitrate may fuse or melt and it may explode. Closed containers may rupture violently due to rapid decomposition, if exposed to fire or excessive heat for a sufficient period of time. During a fire, irritating/toxic nitrogen oxides,

nitrogen and oxygen gases may be generated.

Precautions for Fire-Fighters

Special Protective Equipment and Wear NIOSH-approved self-contained breathing apparatus and protective clothing.

Further Information Not Available

Section 06 - Accidental Release Measures

Equipment / Emergency **Procedures**

Personal Precautions / Protective Wear appropriate personal protective equipment. Ventilate area. Only enter area with PPE. Stop or reduce leak if safe to do so. Flush with water to remove any residue.

Environmental Precautions Prevent material from entering sewers.

Methods and Materials for Containment and Cleaning Up

Clean up spills in a manner that does not disperse dust into the air. Use non-sparking tools and equipment. Reduce airborne dust and prevent scattering by moistening with water. Pick up spill for recovery or disposal and place in a closed container. Small amounts of residue may be flushed to sewer with plenty of water.

Section 07 - Handling and Storage

hygiene and housekeeping practices. Wash thoroughly after handling. Avoid all situations

that could lead to harmful exposure.

Conditions for Safe Storage Keep container dry. Keep in a cool place. Ground all equipment containingmmaterial.

Keep container tightly closed. Keep in a cool and well-ventilated area. Highly toxic or

infectious materials should be stored in a separate locked safety storage cabinet or room.

IncompatibilitiesReacts with acids to emit toxic fumes of nitrogen dioxide. Contact with the following may cause an explosion: barium rhodanide, boron phosphide, cyanides, sodium thiosulfate,

sodium hypophosphite, sulfur plus charcoal, powdered aluminum and aluminum oxide. Fibrous organic material such as jute, wood, and similar cellulosic materials can become

highly combustible by nitrate impregnation.

Section 08 - Exposure Controls and Personal Protection

Exposure Limit(s)

Component Regulation Type of Listing Value

Sodium Nitrate Not Established

Engineering Control(s)

Ventilation Requirements Mechanical ventilation (dilution or local exhaust), process or personnel enclosure and

control of process conditions must be provided in accordance with all fire codes and regulatory requirements. Supply sufficient replacement air to make up for air removed by

exhaust systems.

Other Emergency shower and eyewash must be available and tested in accordance with

regulations and be in close proximity.

Protective Equipment

Eyes/Face Chemical goggles, full-face shield, or a full-face respirator is to be worn at all times when

product is handled. Contact lenses should not be worn; they may contribute to severe eye

injury.

Hand Protection Impervious gloves of chemically resistant material (rubber or PVC) should be worn at all

times. Wash contaminated clothing and dry thoroughly before reuse.

Skin and Body ProtectionBody suite, aprons, and/or coveralls of chemical resistant material should be worn at all

times. Wash contaminated clothing and dry thoroughly before reuse.

Impervious boots of chemically resistant material should be worn at all times. No special

footwear is required other than what is mandated at place of work.

Respiratory Protection For dusty or misty conditions, wear NIOSH-approved dust or mist respirator. In case of

spill or leak resulting in unknown concentration, use NIOSH approved supplied air

respirator.

Thermal Hazards Not Available

Section 09 - Physical and Chemical Properties

Appearance

Physical State Solid

Colour Colourless prills pellets

Odour Odourless

Odour Threshold Not Applicable

Property

pH 5.5-8.3 (5% solution)

Melting Point/Freezing Point 308°C

Initial Boiling Point and Boiling

Range

Decomposes

Flash Point Not Applicable

Evaporation Rate Negligible

Flammability Non-Flammable. Weak oxidizer – can slightly increase the burning rate of a fire.

Upper Flammable Limit Not Applicable

Lower Flammable Limit Not Applicable

Vapour Pressure (mm Hg, 20°C) Extremely low

Vapour Density (Air=1) Not Applicable

Relative Density Not Available

Solubility(ies) Very soluble in water.

Very soluble in ammonia; slightly soluble in ethanol, methanol, acetone and glycerol.

Partition Coefficient: n-

octanol/water

Log Pow = -0.79

Auto-ignition Temperature Not Applicable

Decomposition Temperature Above 510-600°C

Viscosity Not Applicable

Explosive Properties Explosive with shock, heat or friction. Sodium nitrate decomposes explosively when

heated > 538°C.

Specific Gravity (Water=1) 2.26 at 20°C

% Volatiles by Volume 0

Formula NaNO₃

Molecular Weight 84.99

Section 10 - Stability and Reactivity

Reactivity Sodium nitrate is a National Fire Protection Association (NFPA) Class 1 Oxidizer. Class 1

Oxidizers do not moderately increase the burning rate of combustible materials with which

they come into contact.

Stability Product is stable

Possibility of Hazardous

Reactions

None reported.

Conditions to AvoidHeat near decomposing temperature. Contact with incompatible materials. Avoid contact

with flammable or combustible materials.

Incompatible Materials Reacts with acids to emit toxic fumes of nitrogen dioxide. Contact with the following may

cause an explosion: barium rhodanide, boron phosphide, cyanides, sodium thiosulfate, sodium hypophosphite, sulfur plus charcoal, powdered aluminum and aluminum oxide. Fibrous organic material such as jute, wood, and similar cellulosic materials can become

highly combustible by nitrate impregnation.

Hazardous Decomposition

Products

Decomposes appreciably at 510-600 $^{\circ}\text{C}$ forming sodium nitrite, sodium oxide, nitrogen

oxides, nitrogen and oxygen.

Section 11 - Toxicological Information

Acute Toxicity

ComponentOral LD_{50} Dermal LD_{50} Inhalation LC_{50} Sodium Nitrate1267mg/kg (rat)Not AvailableNot Available

Chronic Toxicity – Carcinogenicity

Component IARC

Sodium Nitrate Not known to be carcinogenic.

Skin Corrosion/Irritation Skin irritant

Ingestion Toxic by ingestion. May cause gastroenteritis and abdominal pains. Other symptoms may

include dizziness, bloody diarrhea, convulsions, and collapse. Purging and diuresis can be expected. Small repeated doses may cause headache and mental impairment. Rare cases of nitrates being converted to the more toxic nitrites have been reported, mostly

with infants

Inhalation Inhalation of dust irritates the respiratory tract. Symptoms may include coughing,

shortness of breath.

Serious Eye Damage/Irritation Causes serious eye irritation.

Respiratory or Skin Sensitization Not Available

Germ Cell Mutagenicity

Not known to be mutagenic. May be a mutagenic hazard at very high level doses

according to animal tests.

Reproductive ToxicityMay be a reproductive hazard at very high level doses according to animal tests.

STOT-Single Exposure Irritating to respiratory tract.

STOT-Repeated Exposure May cause methemoglobinemia and anemia. Ingestion of large quantities will cause

cyanosis. Nephritis can result from chronic exposure..

Aspiration Hazard Not Available

Synergistic Materials Not Available

Section 12 – Ecological Information

Ecotoxicity

Component Toxicity to Algae Toxicity to Fish Toxicity to Daphnia and Other Aquatic Invertebrates

Sodium Nitrate Not Available LC_{50} (Orncorhynchus mykiss, LC_{50} (Daphnia magna, 48hr):

96hr): 1050mg/L 323mg/L

Biodegradability Possibly hazardous short-term degradation products are not likely. However, long term

degradation products may arise. The products of degradation are less toxic than the

product itself.

Bioaccumulation Not expected to bioaccumulate.

Mobility Nitrates have a tendency to migrate into groundwater as they do not bind to soil and are

extremely soluble.

Other Adverse Effects Not Available

Section 13 - Disposal Considerations

Waste From Residues/Unused Products

Dispose in accordance with all federal, provincial, and/or local regulations including the

Canadian Environmental Protection Act.

Contaminated Packaging Dispose in accordance with all federal, provincial, and/or local regulations including the

Canadian Environmental Protection Act.

Section 14 – Transport Information

UN Number UN 1498

UN Proper Shipping Name Sodium nitrate

Transport Hazard Class(es) 5.1

Packaging Group

Environmental Hazards Not listed as a marine pollutant under Canadian TDG Regulations, schedule III.

Special Precautions Not Available

Transport in Bulk Not Available

Additional Information Packing Group Limited Quantity Index

5 Kg

<u>TDG</u>

Other Secure containers (full and/or empty) with suitable hold down devises during shipment and

ensure all caps, valves, or closures are secured in the closed position.

TDG PRODUCT CLASSIFICATION: This product has been classified on the preparation date specified at section 14 of this MSDS / SDS, for transportation in accordance with the requirements of part 2 of the Transportation of Dangerous Goods Regulations. If applicable, testing and/or published test data regarding the classification of this product are listed in the references at section 16 of this MSDS / SDS.

Section 15 – Regulatory Information

NOTE: THE PRODUCT LISTED ON THIS SDS HAS BEEN CLASSIFIED IN ACCORDANCE WITH THE HAZARD CRITERIA OF THE CANADIAN CONTROLLED PRODUCTS REGULATIONS. THIS SDS CONTAINS ALL INFORMATION REQUIRED BY THOSE REGULATIONS.

Section 16 - Other Information

Preparation Date

August 12, 2015

Note: The responsibility to provide a safe workplace remains with the user. The user should consider the health hazards and safety information contained herein as a guide and should take those precautions required in an individual operation to instruct employees and develop work practice procedures for a safe work environment. The information contained herein is, to the best of our knowledge and belief, accurate. However, since the conditions of handling and use are beyond our control, we make no guarantee of results, and assume no liability for damages incurred by the use of this material. It is the responsibility of the user to comply with all applicable laws and regulations.

Attention: Receiver of the chemical goods / SDS coordinator

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If you have any questions or concerns please call our customer service center.

References:

- 1) CHEMINFO
- 2) eChemPortal
- 3) TOXNET
- 4) Transportation of Dangerous Goods Canada
- 5) HSDB
- 6) ECHA
- 7) PAN

ClearTech Industries Inc. - Locations

Corporate Head Office: 1500 Quebec Avenue, Saskatoon, SK, S7K 1V7

Phone: 1(306) 664 – 2522

Fax: 1(888) 281-8109

www.cleartech.ca

24 Hour Emergency Number - All Locations - 1(306) 664-2522